#### Standards for proton and heavy ion beams

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#### Outline -

#### • Traceability chain for proton and ion therapy beams

- Current and projected particle facilities
- Traceability chain of protons and ions compared to Co-60

#### • Primary standards for absorbed dose

- Physical basis of operation
- Example development efforts

#### • Effect on end-user dosimetry in the next 5-10 years

- Implications for protocols
- How dissemination might work
- Reductions in measurement uncertainty

#### Treatment modalities available in the clinic -

#### http://www.ptcog.ch



#### Summary as of 01-April-2015

#### p, C<sup>+</sup>, both

Location	Active	Construction
US	16	14
World	<b>45, 4</b> , 4	29, 2, 2

# What is 'traceability'?

- Traceability is a property of a dose measurement! It ensures that the measurement can be related to the national dose standard "D<sub>w</sub>" maintained by a PSDL through a documented unbroken chain of calibrations, each contributing to the measurement uncertainty.
  - **PSDL**: Primary Standards Dosimetry Laboratory
    - e.g. NIST, NRCC, ...
- The national dose standard Dw is realized at the PSDL using a primary instrument.
  - **Primary instrument:** for direct realization of absorbed dose
    - e.g. calorimeter, Fricke solution, ionization chamber, Faraday cup, ...

# What is 'traceability'? (continued)

- The relationship (or cross calibration) between the measurement and the national standard for dose, Dw, is done by following accepted **protocols**.
  - **protocol**: prescribes reference beam conditions, procedures for conducting and correcting measurements, ...
    - e.g. AAPM TG-51, IAEA TRS-398, ...
- Traceability can be direct or indirect ...





#### Dosimeter calibration for p and light-ion beams -

- Need an appropriate absolute dosimeter<sup>1</sup>
  - Faraday cups used in the early days
  - Other options include: Fricke dosimetry, ionization chambers, carbon activation, calorimetry
  - ICRU 78 (2007)  $\rightarrow$  calorimetry, when available...





HUPTI, 2012 – photo credit: M.A. Moyers

<sup>1</sup>Moyers, M.A. and Vatnitsky, S.M., "Practical Implementation of Light Ion Beam Treatments", Medical Physics Publishing, Madison, WI, 2012, p. 24.

#### Water calorimetry – basis of operation





#### Water calorimetry corrections: heat transfer (ht)



Modeling the effect with finite-element analysis is straightforward, enabling the determination of a correction factor  $\mathbf{k}_{ht}$ . The cumulative effect of thermal gradients within the device distorts the waveform (drift segments not parallel) and introduces a systematic uncertainty into  $\Delta T$ .



#### Water calorimetry corrections: heat defect (HD)

A much larger systematic effect is attributable to photolytic reactions induced by radiation.

Table 1. Model IIIR: reactions and rate constants (4 °C)

Reactions*				Rate constants <sup>b</sup>	
1	$e_{aq}^- + e_{aq}^-$	$\rightarrow$	$H_2 + OH^- + OH^-$	$3.48 \times 10^{9}$	
2	em + H	$\rightarrow$	$H_2 + OH^-$	$1.73 \times 10^{10}$	
3	$e_{aq} + OH$	$\rightarrow$	OH-	$2.38 \times 10^{10}$	
4	$e_{aq} + H_2O_2$	$\rightarrow$	OH + OH	$8.84 \times 10^{2}$	
5	eaq +O2	$\rightarrow$	O2	$1.16 \times 10^{10}$	
6	e-a + 02	$\rightarrow$	HO <sub>2</sub> + OH <sup>-</sup>	$8.48 \times 10^{9}$	
7	$e_{aq}^- + HO_2$	$\rightarrow$	HO	$8.48 \times 10^{9}$	
8	H + H	$\rightarrow$	H <sub>2</sub>	$3.44 \times 10^{9}$	
9	H + OH	$\rightarrow$	H <sub>2</sub> O	$1.21 \times 10^{10}$	
10	$H + H_2O_2$	$\rightarrow$	$OH + H_2O$	$3.18 \times 10^{7}$	
11	$H + O_2$	$\rightarrow$	HO <sub>2</sub>	$9.58 \times 10^{9}$	
12	$H + HO_2$	$\rightarrow$	H <sub>2</sub> O <sub>2</sub>	$7.24 \times 10^{9}$	
13	$H + O_2$	$\rightarrow$	HO <sub>2</sub>	$7.24 \times 10^{9}$	
14	OH + OH	$\rightarrow$	H <sub>2</sub> O <sub>2</sub>	$3.76 \times 10^{9}$	
15	OH + H <sub>2</sub>	$\rightarrow$	$H + H_2O$	$2.40 \times 10^{7}$	
16	$OH + H_2O_2$	$\rightarrow$	$H_2O + H_2O$	$1.79 \times 10^{7}$	
17	OH + HO <sub>2</sub>	$\rightarrow$	$H_2O + O_2$	$9.08 \times 10^{9}$	
18	OH + 02	$\rightarrow$	$OH^- + O_2$	$7.89 \times 10^{9}$	
19	HO <sub>2</sub> + HO <sub>2</sub>	$\rightarrow$	$H_2O_2 + O_2$	$3.72 \times 10^{5}$	
20	HO2 + O2	$\rightarrow$	$H_2O_2 + O_2 + OH^-$	$5.84 \times 10^{7}$	
21	H <sub>2</sub> O	$\rightarrow$	H* + OH-	$2.22 \times 10^{-6}$	
22	H* + OH-	$\rightarrow$	H <sub>2</sub> O	7.23 × 1010	
23	H <sub>2</sub> O <sub>2</sub>	$\rightarrow$	H* + HO2	$1.34 \times 10^{-2}$	
24	$H^* + HO_2$	$\rightarrow$	H <sub>2</sub> O <sub>2</sub>	$3.13 \times 10^{10}$	
25	$H_2O_2 + OH^-$	$\rightarrow$	$HO_2 + H_2O$	$7.56 \times 10^{9}$	
26	HO <sub>2</sub> + H <sub>2</sub> O	$\rightarrow$	$H_2O_2 + OH^-$	$5.45 \times 10^{3}$	
27	н	$\rightarrow$	em + H*	$8.83 \times 10^{-1}$	
28	e_ + H*	$\rightarrow$	H	$1.88 \times 10^{20}$	
29	e_ + H_O	$\rightarrow$	$H + OH^-$	$5.08 \times 10^{\circ}$	
30	H + OH-	$\rightarrow$	em + H2O	$7.77 \times 10^{6}$	
31	OH	$\rightarrow$	H* + O*	$1.34 \times 10^{-2}$	
32	H* + O"	$\rightarrow$	OH	$3.13 \times 10^{10}$	
33	OH + OH	$\rightarrow$	$O^- + H_2O$	$7.56 \times 10^{9}$	
34	0" + H2O	$\rightarrow$	OH-+OH	$5.45 \times 10^{3}$	
35	HO <sub>2</sub>	$\rightarrow$	O5 + H*	$4.21 \times 10^{5}$	
36	O; + H*	$\rightarrow$	HO <sub>5</sub>	$3.13 \times 10^{10}$	
37	HO2 + OH-	$\rightarrow$	05 + H <sub>2</sub> O	$7.91 \times 10^{9}$	
38	05 + H-O	$\rightarrow$	HO <sub>2</sub> + OH <sup>-</sup>	$1.94 \times 10^{-2}$	
39	0" + H2	$\rightarrow$	H + OH	$7.95 \times 10^{7}$	
40	0-+H-0-	$\rightarrow$	05 + H-O	$3.44 \times 10^{8}$	
41	OH + HO;	$\rightarrow$	OH + HO,	$5.17 \times 10^{9}$	
42	OH + O-	$\rightarrow$	HOS	$6.02 \times 10^{9}$	
43	e- + HO5	$\rightarrow$	0" + OH-	$2.19 \times 10^{9}$	
44	e_ +0-	$\rightarrow$	OH- + OH-	$1.82 \times 10^{10}$	
45	0+0-	-	05	$2.63 \times 10^{9}$	
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Computational modeling of this effect also is straightforward, but requires adequate models of the reaction system (e.g. at left) and G-values (average photolytic production yield, #/100 eV) for production of reactant species by the radiation beam. The associated correction factor is designated  $\mathbf{k}_{HD}$ .

N.V. Klassen and Carl K. Ross, J. Res. Natl. Inst. Stand. Techol. **107**, 171-178 (2002).

#### Water calorimetry overview cont'd ...

$$D_{\rm w} = \Delta T_{\rm w} \cdot c_{\rm w} \cdot k_{\rm HD} \cdot k_{\rm ht} \cdot k_{\rm dd} \cdot k_{\rm p}$$

- $k_{\text{HD}}$  correction for the heat defect, h, due to radiation-induced chemical reactions, where  $k_{\text{HD}} = (1-h)^{-1}$ . It is considered here to be unity, with an appropriate uncertainty;
- correction for heat transfer due to thermal gradients generated within the vessel;
- k<sub>dd</sub> correction for the non-uniformity of the lateral dose profile between the thermistor beads and the centre of the field;
- k<sub>p</sub> correction of the perturbation of the radiation field due to attenuation and scattering by the thermistor probes and glass vessel, determined from ionization chamber measurements and Monte Carlo simulations.

#### NIST at HUPTI (Hampton, VA, 2012)

With the NIST calorimeter in place, HUPTI generated a treatment plan with a 10 cm by 10 cm field, with a range of 16 cm and a modulation of 10 cm. This placed the calorimeter thermistor probes inside the vessel at an 11 cm depth, precisely in the middle of the uniform dose region. Chambers were positioned at the same depth for comparison measurements.



#### NIST/HUPTI – 2012, cont'd...



#### McGill University/MGH - 2010 (Sarfehnia, Seuntjens - McGill)

Direct absorbed dose to water determination based on water calorimetry in scanning proton beam delivery



FIG. 2. The consol, temperature distribution results inside a geometrical model of our setup. Only one quarter of the entire geometry has been modeled due to symmetry. A picture of the parallel plate vessel (with two thermistors positioned inside) is also included.



Fig. 1. A schematic diagram of the McGill in-house built transportable water calorimeter positioned below a proton gantry.

TABLE V. The final dose measurement results and comparison between the primary water calorimetry and reference T1 mini-Shonka (based on a pulsed beam ion recombination criterion).

	$k_{ m ht}$	Calorimetry (Gy/MU)	T1 Chamber (Gy/MU)	% difference
Scattering Scanning	0.996 0.953	$\begin{array}{c} 9.087 \times 10^{-3} \\ 1.198 \times 10^{-3} \end{array}$	$\begin{array}{c} 9.118 \times 10^{-3} \\ 1.203 \times 10^{-3} \end{array}$	0.34 0.42

Medical Physics, Vol. 37, No. 7, July 2010

#### METAS/PSI - 2006

Water calorimetry with scanned proton beam, showing response of each thermistor



Figure 1: The sealed vessel of the water calorimeter with two thermistors separated by 1 cm. Superimposed is a schematic scanned proton dose profile.



Figure 3: The temperature rise signal of the two thermistors for the cubic dose distibution of 4.5 Gy.

http://metascms01.admin.ch/metasweb/Fachbereiche/Ionisierende Strahlung und Radioaktivitaet/IS%20PDF%20Dateien/P rotonkalorimeter METAS PSI Jahresbericht.pdf Work of Gagnebin et al.

#### NPL – 2004 (to present)



Figure 3. Schematic diagram of the small-body portable graphite calorimeter (left, not to scale, adapted from McEwen and Duane (2000)) and cross sections of the core and the jacket (right, approximately to scale).

http://metascms01.admin.ch/metasweb/Fachbereiche/Ionisierende St rahlung und Radioaktivitaet/IS%20PDF%20Dateien/Protonkalorimeter METAS PSI Jahresbericht.pdf Work of Palmans et al.



Design of the graphite calorimeter for primary dosimetry in proton and ion beams

30 mm

30 mm

#### http://www.npl.co.uk/news/from-graphite-to-water



General views of the assembled electron/photon therapy level absorbed-dose graphite calorimeter (foreground), and the proton and light ion calorimeter, now being commissioned

http://www.npl.co.uk/upload/pdf/graphitecalorimeters-absorbed-dose.pdf

## Graphite vs. water calorimetry

- Requires conversion from dose-to-graphite to dose-towater
- Heat capacity of graphite is lower ( $\Delta T/Gy$  higher)
- Choice of passive (quasi-adiabatic) or active (isothermal) modes of operation
- Heat conductivity of graphite is higher
- Absorber medium is a solid (no buoyant convection)
- Heat defect involves crystal dislocations

## Effect on end-user dosimetry in 5-10 years

#### *If...*

- Protocols are worked out
  - beam quality specifiers (e.g. reference depths, ...)?
  - Scattered and scanned?

#### Dissemination difficulties addressed

 Onsite calibrations by NMIs without calibration ranges (e.g. along the lines of the BIPM K6 Key Comparison?)

Then...

#### Reductions in measurement uncertainty

- Combined standard uncertainty for  $N_{D,w,Co-60} = 0.47\%$ <sup>1</sup>
- " for  $k_Q$  in proton beams = 1.7% <sup>2</sup>
- " for  $k_Q$  for ion beams = 2.8%  $^3$

<sup>1</sup>NIST Special Publication 250-74, p. <u>http://nist.gov/calibrations/upload/sp250-74.pdf</u> <sup>2,3</sup>IAEA Technical Report Series No. 398, pp. 194, 197, respectively, http://wwwpub.iaea.org/mtcd/publications/pdf/trs398\_scr.pdf

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- Michael Mitch, PhD

# Primary dose standards for therapy beams of protons ...

<mark>4%</mark>	a.	cannot be realized within the same instrument that would be used for C <sup>+</sup> ions.
18%	b.	will be most beneficial for the determination of chamber $k_Q$ factors.
7%	C.	should be built with a Faraday cup, because it is the best established technique.
11%	d.	will be built with graphite because of its relatively low heat defect.
61%	e.	will likely involve calorimetry, according to recommendations in ICRU Report 78.

# Primary dose standards for therapy beams of protons ...

1. Primary dose standards for therapy beams of protons ...

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- (d) will be built with graphite because of its relatively low heat defect.
- (e) will likely involve calorimetry, according to recommendations in ICRU Report 78.

Answer: e - will likely involve calorimetry, according to recommendations in ICRU Report 78.

Ref: Moyers, M.A. and Vatnitsky, S.M., "Practical Implementation of Light Ion Beam Treatments", Medical Physics Publishing, Madison, WI, 2012, p. 24.

Within the US, traceability to national standards for dosimetry of proton and ion beams ...

37%	a.	requires the commissioning of a national primary standard for proton absorbed dose.
46%	b.	necessarily involves Co-60 as the reference beam quality.
14%	C.	awaits acquisition of suitable particle beam facilities by NIST.

0% d. is feasible for double-scattered beams only.

e. entails formal recognition by the BIPM.

3%

# Within the US, traceability to national standards for dosimetry of proton and ion beams ...

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- (b) necessarily involves Co-60 as the reference beam quality.
- (c) awaits acquisition of suitable particle beam facilities by NIST.
- (d) is feasible for double-scattered beams only.
- (e) entails formal recognition by the BIPM.

Answer: b – necessarily involves Co-60 as the reference beam quality.

Ref: Moyers, M.A. and Vatnitsky, S.M., "Practical Implementation of Light Ion Beam Treatments", Medical Physics Publishing, Madison, WI, 2012, p. 29.

Ref: "Absorbed Dose Determination in External Beam Radiotherapy: An International Code of Practice for Dosimetry Based on Standards of Absorbed Dose to Water, IAEA Technical Report Series TRS-398 (2000), p. 135, 151.